

Extent of Nitrate and Nitrite Pollution in Ground Water of Rural Areas of Lucknow, U.P., India

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<http://dx.doi.org/10.12944/CWE.9.1.17>

(Received: December 08, 2013; Accepted: January 17, 2014)

ABSTRACT

The present world is facing problems with a wide variety of pollutants. Water pollution is a major global problem. It has been suggested that it is the leading worldwide cause of deaths and diseases and that it accounts for the deaths of more than 14,000 people daily in Lucknow Capital of Uttar Pradesh in India. Nitrate and Nitrite pollution is one of groundwater's most commonly identified contaminants, an indicator of serious pollution as they are associated with septic waste and agricultural endeavours, leads to numerous health problems to human beings and animals. 4 rural areas of Lucknow were selected and 15 samples from each station to check the level of nitrite and nitrate parameters in groundwater. Further our studies reveal that the extent of nitrate and nitrite varied with reference to sampled site and maximum nitrate was found to be 250.224 and maximum nitrite was 1.8998 both are high.

Key words: Nitrite, Nitrate, Ground Water Pollution, Lucknow.

INTRODUCTION

Healthy soil, clean water and air are the soul of life. Soil, water and air are no longer clean and pure, today pose human health risks. Comprising over 70% of the Earth's surface water is undoubtedly the most precious natural resource that exists on our planet. It is essential for everything on our planet to grow and prosper. Water pollution remains one of the most visible and persistent signs of our impact on the natural world. Gomti river in Lucknow city in India, receives huge quantities of untreated waste, from industrial effluents to domestic discharge, the river becomes more of a flowing dumping yard for the 15 smaller and bigger towns in its catchment area which affects badly on human health. Although we as humans recognize this fact we disregard it by polluting our rivers, lakes, and oceans. The water pollutants include sewage, variety of both organic and inorganic pollutants including oils, greases, plastics plasticizers, metallic wastes, suspended solids, phenols, acids, greases, salts, dyes, cyanides, DDT and some heavy metals

like Cu, Cr, Cd, Hg, Pb are also discharged from industries¹. The contamination of the environment with toxic metals has become a worldwide problem, affecting crop yields, soil biomass and fertility². In Lucknow Gomti river collects large amounts of human and industrial pollutants as it flows through the highly populous areas (18 million approx) of Uttar Pradesh. High pollution levels in the river have negative effects on the ecosystem of the Gomti threatening its aquatic life and also surrounded areas of Lucknow. All industries of distillery, milk industry and vegetable oil, pouring effluent directly into Gomti and besides this domestic waste water are also discharge into the River Gomti. Drains are the main source of water pollution especially for rivers flowing within the city carry industrial effluent, domestic waste, sewage, and Medicinal waste results in pouring the water quality³. The specific contaminants leading to pollution in water include a wide spectrum of chemicals, pathogens and physical or sensory changes such as elevated temperature and discoloration. While many of the chemicals and substances that are regulated may

be naturally occurring (calcium, sodium, iron, manganese, etc.) the concentration is often the key in determining what is a natural component of water, and what is a contaminant. High concentrations of naturally occurring substances can have negative impacts on aquatic flora and fauna. Water pollution can cause by both organic and inorganic pollutants. Nitrate is an inorganic compound that can be a natural or man made contaminant in drinking water. Nitrite and Nitrate pollution is due to excessive amount of nitrate in surface or ground water as a result of agricultural practices. Farmers and home owners using nitrate bearing fertilizers often use a variety of pesticides and herbicides which may migrate to ground water supplies. Due to its high solubility in water, nitrate and nitrite are the most common contaminants in rural and suburban areas. Fertilizer use has led to greater contamination of surface and groundwater with nitrates essentially dissolved nitrogen fertilizer that has not been taken up by plants. Nitrate (NO_3) is the main form in which nitrogen occurs in groundwater, although dissolved nitrogen may also be present as nitrite (NO_2), ammonium (NH_4), nitrous oxide (N_2O) and organic nitrogen⁴. Nitrate and Nitrite are the inorganic pollutants which degrades the water quality of drinking water. Higher concentration of metal in water and could be due to the industrial, agricultural or domestic runoff coming into the river⁵. River water quality monitoring is necessary especially where the water serves as drinking water sources⁶. Nitrate and Nitrite are the inorganic pollutants which degrades the water quality of drinking water. Although there are many sources of nitrogen (both natural and anthropogenic) that could potentially lead to the pollution of the groundwater with nitrates, the anthropogenic sources are really the ones that most often cause the amount of nitrate to rise to a dangerous level. Waste materials are one of the anthropogenic sources of nitrate contamination of groundwater. Water moving down through soil after rainfall or irrigation carries dissolved nitrate with it to ground water. In this way, nitrate enters the water supplies of many home owners who use wells or springs. Many areas of the United States and other countries have reported significant contamination of groundwater from septic tanks. Ground water contamination is usually related to the density of septic systems⁷. Nitrogen in organic form and ammonium can be converted by bacteria in aerobic

conditions into nitrite and nitrate, a process termed 'nitrification'. Nitrate in anaerobic systems can be reduced by other strains of bacteria to nitrous oxide or nitrogen gas, by 'denitrification'. In aerobic water nitrogen occurs as nitrate or nitrite ions. Nitrate is stable over a considerable range of conditions and is very mobile in water. Ammonium and organic forms are unstable and are generally considered to be indicators of pollution. Drinking water high in nitrate is potentially harmful to human and animal health. Nitrate (NO_3) is a naturally occurring form of nitrogen (N) which is very mobile in water⁸. Nitrate pollution for groundwater supplies is directly related to the amount of fertilizers or other nitrogen inputs to the land, as well as the permeability of the soil. In China assessment of groundwater contamination happened by nitrates associated with wastewater irrigation.⁹ The United States Environmental Protection Agency is currently establishing National Primary Drinking Water Regulations for over 80 contaminants under the Safe Drinking Water Act and to reduce the contaminant concentrations of all drinking water to levels near those prescribed in the Maximum Contaminant Level Goals¹⁰. Comprehensive assessment of Freshwater Water Resources and water availability in the world was done.¹¹ Effect of nitrate on drinking water quality and its management¹².¹³ Northern China affected by Nitrate Pollution in Groundwater. Nitrate in drinking water can be effectively reduced in a number of ways. The best solution is to find an alternative water supply for drinking and cooking purposes. If other pollutants are not present, reverse osmosis systems, anion exchange units, and distillation can reduce nitrate and nitrite levels. Objectives of the study were to determine the extent of nitrate and nitrite concentration in ground water of some areas of Lucknow. Four different stations of rural areas were selected namely Raibareilly Road, Kanpur Road, Sultanpur Road and Hardoi Road. 15 samples were collected namely from each station.

Sample Collection

The sampling of ground water was done from 4 different stations of rural areas of Lucknow. 15 samples from each station were taken. All the samples were taken from deep well hand pumps. Each sampling station covers nearly 16 Km area. Name of sites from different station are as follows:

Sampling Station (Raibareli Road)

1.Kudha, 2.Merai Khera, 3.Atrauli, 4.Kankaha Gaon, 6.Gadiyana, 7.Sikandarpur, 8.Katua Khera, 9.Kesari Khera, 10.Madhav Khera, 11.Harkansh Khera, 12.Pachauri, 13. Hualas Khera,14. Ranjeet Khera, 15. Kankaha Bazar, 16. Badan Kher.

Sampling Station (Kanpur Road)

1.Narayan Khera 2.Hindu Khera 3. Banthara Bazar 4.Kati Bagia 5. Piparsand 6. MunnaKhera 7.SaraiSahjadi 8. BalluKhera 9. Bauri Khera 10. Balhe Mau 11. Nidhaan Khera 12.Gauri 13. ShivPura 14. Shaikhpur 15. Bakhat Khera.

Sampling Station (Sultanpur Road)

1. Kasimpur Biruha 2.Gusaiganj 3.Pancham Purwa 4.Kasimpur 5. NawabAliPurwa 6.Amirpur 7.Begaria 8.Sengta 9.Kabirpur 10.Bikauli 11.Pahar Nagar 12.Malauli 13.Hardaspur 14.Salauli 15. Jahangirpur.

Sampling Station (Hardoi Road)

1.Suspan 2.Thari 3.Gahdon 4. Dilawar Nagar 5.Rahimabad 6. Jamoliya 7. Kiatholiya 18. Gopalpur9. Mundiyyara 10. Mahima Khera 11. Badkhorwa 12.Kamaaluddin Nagar 13.Ater 14.Malihabad 15. Maal.

MATERIALS AND METHOD

Samples were collected in precleaned bottles and labelled at the site. All samples were analysed for nitrate and nitrite concentrations within 24 hours of sampling to minimise the effect of storage by freezing and to obtain more reliable results. Presence of nitrate and nitrite are normally observed by yellow and pink colour intensity produced by Salicylic acid and NED –N-diamine dihydrochloride 1Naphthylethylene.¹⁴Rapid

Dilution of Stock Solution

	Stock Sol. (ml)	Dist. Water (ml)	Concentration
1.	0.2	9.8	20
2.	0.4	9.6	40
3.	0.6	9.4	60
4.	0.8	9.2	80
5.	1.00	9.00	100

colorimetric determination of nitrate in plant tissues. Estimation of Nitrite content in soil and leaves.

Reagents: For Nitrate**5% Salicylic acid**

Dissolve 5 gm of Salicylic acid in 100 ml of conc. H₂SO₄ or 1.25 gm of salicylic acid in 25 ml of conc. H₂SO₄.

2 N NaOH Solution

Dissolve 40 gm of NaOH Pellets in 500 ml in distilled water.

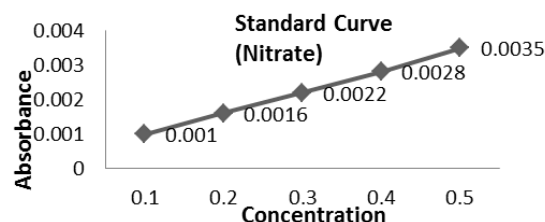
Preparation of Standard Curve for Nitrate Solution

Dissolve 0.1 gm of KNO₃ sail in 100 ml of distilled water.

Procedure

Water sample 0.1 m, 0.4 ml Salicylic acid, 9.5 ml 2N NaOH Orange/Yellow colour intensity indicates the presence of nitrate in water sample. Ing noted at 410 nm by Cary Varian Bio-Spectrophotometer

	Concentration (mg/l)	Absorbance (Optical Density)
1.	20	0.0266
2.	40	0.0456
3.	60	0.0659
4.	80	0.0829
5.	100	0.1101

**Reagents: For Nitrite**

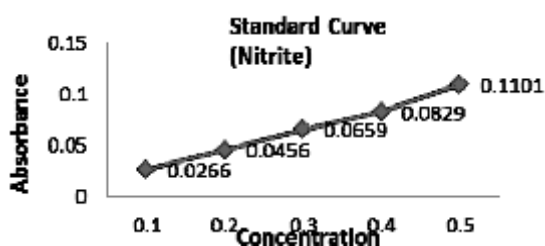
0.01% NED- N-1-Naphthylethylene diamine dihydrochloride 0.01 g in 100 ml of distilled water, 0.02 % sulphanilamide in N HCl.

Preparation of Standard Curve for Nitrite: Stock Solution

0.00g of NANO₂ in 100 ml distilled water

Dilution of Stock Solution

	Stock Sol. (ml)	Dist. Water (ml)	Concentration	Concentration (mg/l)	Absorbance (Optical Density)
1.	0.1	9.9	0.1	0.1	0.0010
2.	0.2	9.8	0.2	0.2	0.0016
3.	0.3	9.7	0.3	0.3	0.0022
4.	0.4	9.6	0.4	0.4	0.0028
5.	0.5	9.5	0.5	0.5	0.0035



1.	Kudha	10.712
2.	Merai Kuera	7.592
3.	Atrauli	1.456
4.	Kankaha Gaon	13.52
5.	Gadiyana	22.568
6.	Sikandarpur	17.784
7.	Katua khera	14.248
8.	Kesari khera	0.624
9.	Madhav khera	14.248
10.	Harkansh khera	32.448
11.	Panchauri	43.784
12.	Hulas khera	1.352
13.	Ranjeet khera	11.336
14.	Kankaha khera	3.432
15.	Badan khera	11.752

Procedure

Water sample 1 ml, 1 ml NED, 1 ml Sulphanilamide. Pink colour intensity indicates the presence of nitrite in water sample. Read it at 540 nm by Cary Varian Bio-Spectrophotometer

Calculation of K- Factor (Nitrate)

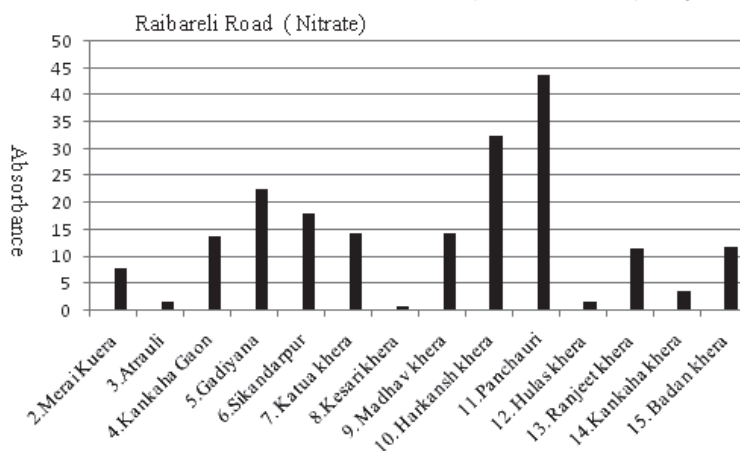
$K1 = 20/0.0266 = 751.87$, $K2=40/0.0456=877.19$, $K3=60/0.0659= 910.47$, $K4= 80/0.0829= 965.01$, $K5= 100/0.1101= 908.26$, (K. Aver = $4412.80/5 = 882.56$).

Calculation of K- Factor (Nitrite)

$K1 = 0.1/0.0010 = 100$, $K2=0.2/0.0016=125$, $K3=0.3/0.0022= 136.6$, $K4=0.4 / 0.0028= 142.85$, $K5=0.5/0.0035= 142.85$. (K. Avery = $647.06/5 = 129.4$).

RESULTS AND DISCUSSION

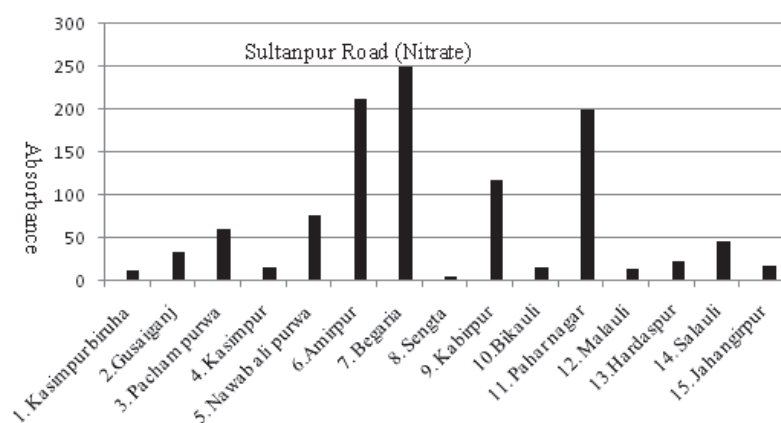
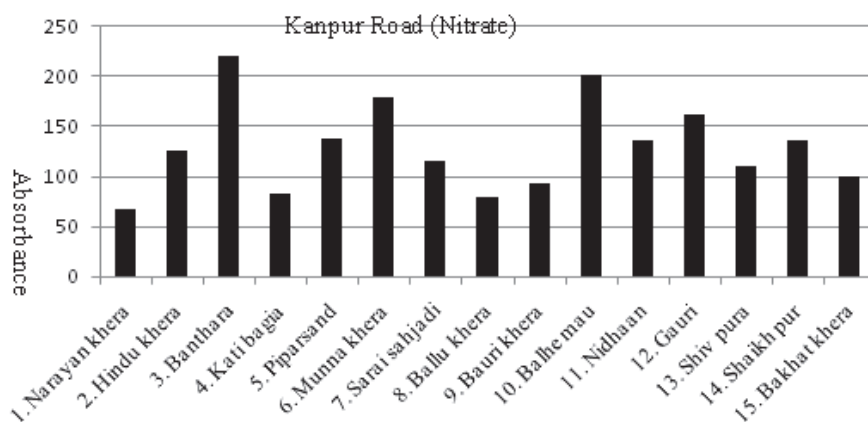
Water contamination caused by the presence of excessive amounts of nitrates washed out from inorganic fertilizers. The presence in water of harmful or objectionable material in sufficient quantity to measurably degrade water quality. The



largest anthropogenic sources are septic tanks, application of nitrogen-rich fertilizers and agricultural processes. Common sources of nitrate include fertilizers and manure, animal feedlots, municipal wastewater and sludge, septic systems,

and N-fixation from atmosphere by legumes, bacteria and lightning.

The maximum level of Nitrate and Nitrite determined in ground water are found to be 250.224

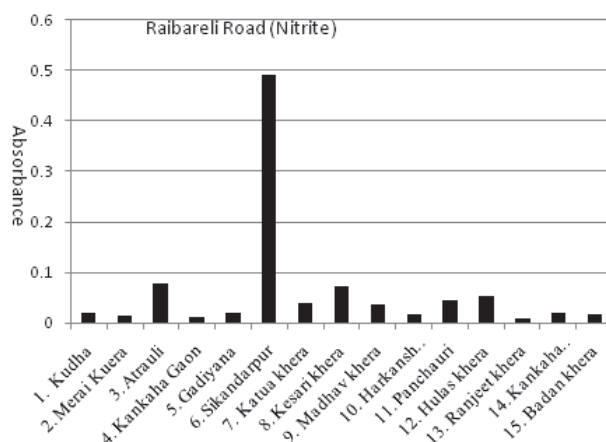
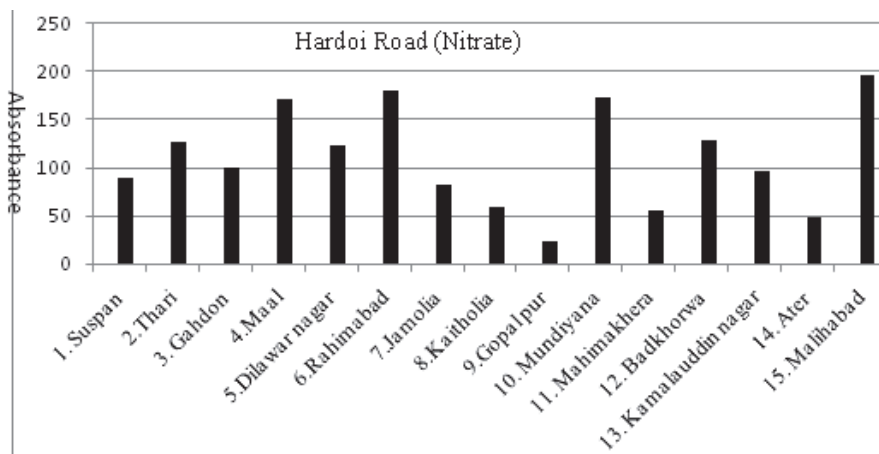


1.	Narayan khera	66.456	1.	Kasimpur biruha	12.896
2.	Hindu khera	126.568	2.	Gusaiganj	33.488
3.	Banthara	221.416	3.	Pancham purwa	61.152
4.	Kati bagi	83.616	4.	Kasimpur	16.64
5.	Piparsand	138.112	5.	Nawabali purwa	76.856
6.	Munna khera	179.504	6.	Amirpur	212.68
7.	Sarai sahjadi	115.96	7.	Begaria	250.224
8.	Ballu khera	78.52	8.	Sengta	4.888
9.	Bauri khera	93.392	9.	Kabirpur	117.832
10.	Balhe mau	201.968	10.	Bikauli	15.288
11.	Nidhaan	135.824	11.	Pahar nagar	200.616
12.	Gauri	162.448	12.	Malauli	15.08
13.	Shiv pura	110.968	13.	Hardaspur	22.984
14.	Shaikh pur	136.968	14.	Salauli	47.32
15.	Bakhat khera	100.152	15.	Jahangirpur	18.512

mg/L (Begaria region) in Sultanpur road and 1.899 mg/L (Banthara region) in Kanpur road respectively. The enhanced levels of Nitrate and Nitrite may be due to excessive application of fertilizer, manures and irrigation.

CONCLUSION

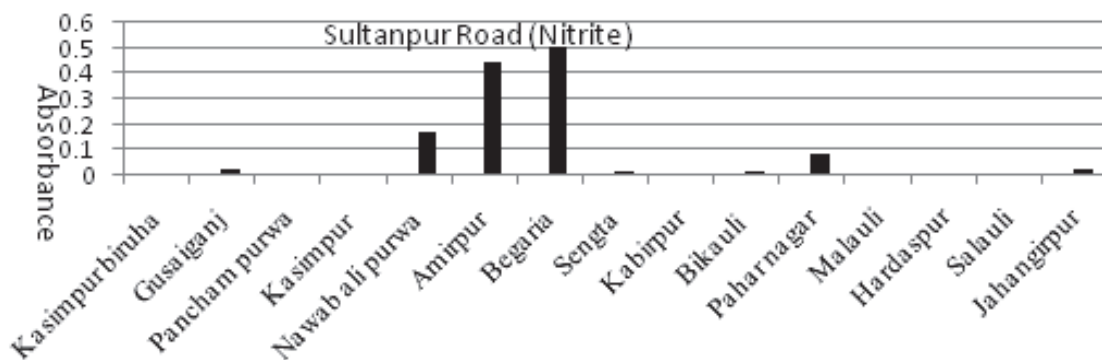
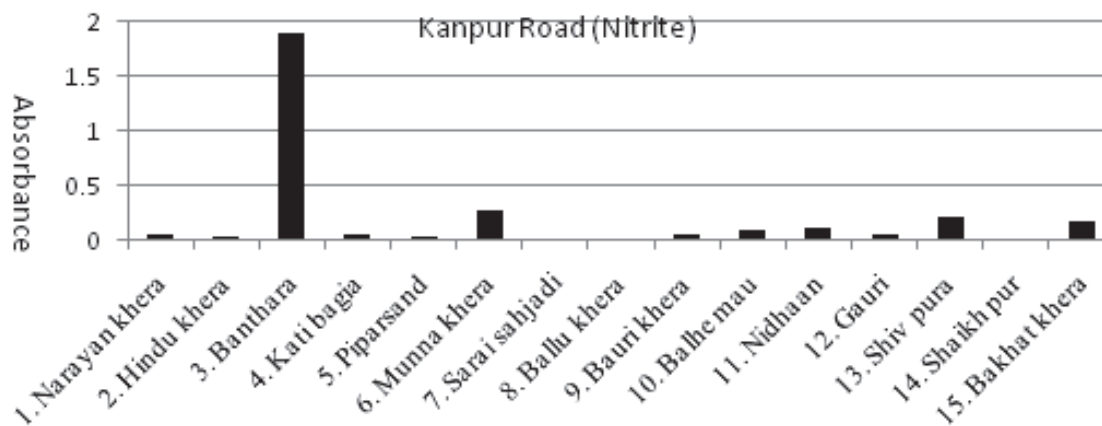
The problems associated with water pollution have the capabilities to disrupt life on our planet to a great extent. No. of organizations



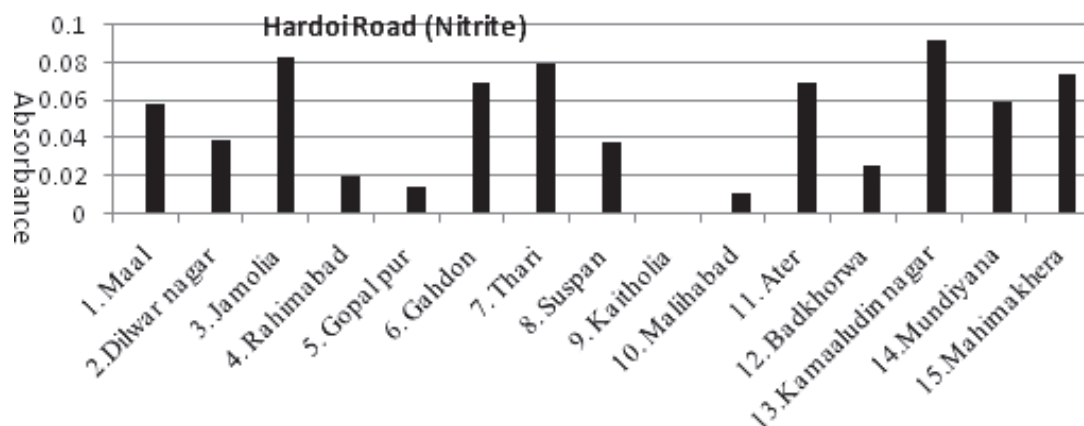
1.	Suspan	89.752	1.	Kudha	0.0204
2.	Thari	126.464	2.	Merai Kuera	0.0137
3.	Gahdon	100.568	3.	Atrauli	0.0785
4.	Maal	171.808	4.	Kankaha Gaon	0.0121
5.	Dilwar nagar	123.344	5.	Gadiyana	0.0198
6.	Rahimabad	180.128	6.	Sikandarpur	0.492
7.	Jamolia	82.056	7.	Katua khera	0.039
8.	Kaitholia	59.8	8.	Kesari khera	0.072
9.	Gopal pur	22.36	9.	Madhav khera	0.0366
10.	Mundiyan	173.888	10.	Harkansh khera	0.0159
11.	Mahimakhera	54.704	11.	Panchauri	0.0433
12.	Badkhorwa	127.712	12.	Hulas khera	0.0516
13.	Kamalauddin nagar	96.616	13.	Ranjeet khera	0.0084
14.	Ater	48.984	14.	Kankaha khera	0.0182
15.	Malihabad	196.872	15.	Badan khera	0.0162

including governmental and non-governmental are trying to combat water pollution thus acknowledging the fact that water pollution is, indeed a serious issue. But the government alone cannot solve the entire problem. It is ultimately up to us, to be informed, responsible and involved when it comes to the problems we face with our water. We must

become familiar with our local water resources and learn about ways for disposing harmful household wastes so they do not end up in sewage treatment plants that cannot handle them or landfills not designed to receive hazardous materials. In our agricultural fields, we must determine whether additional nutrients are needed before fertilizers



1.	Narayan khera	0.0412	1.	Kasimpur biruha	0.0029
2.	Hindu khera	0.0244	2.	Gusaiganj	0.0195
3.	Banthara	1.8998	3.	Pancham purwa	0.0027
4.	Kati bagia	0.0384	4.	Kasimpur	0.0037
5.	Piparsand	0.0186	5.	Nawabali purwa	0.1639
6.	Munna khera	0.2583	6.	Amirpur	0.4413
7.	Sarai sahjadi	0.0113	7.	Begaria	0.502
8.	Ballu khera	0.0011	8.	Sengta	0.0093
9.	Bauri khera	0.0352	9.	Kabirpur	0.0037
10.	Balhe mau	0.0831	10.	Bikauli	0.0067
11.	Nidhaan	0.0988	11.	Pahar nagar	0.0803
12.	Gauri	0.0393	12.	Malauli	0.002
13.	Shiv pura	0.2063	13.	Hardaspur	0.003
14.	Shaikh pur	0.0089	14.	Salauli	0.0038
15.	Bakhat khera	0.1527	15.	Jahangirpur	0.0175



1.	Maal	0.0582
2.	Dilwar nagar	0.0389
3.	Jamolia	0.083
4.	Rahimabad	0.0203
5.	Gopal pur	0.0147
6.	Gahdon	0.0697
7.	Thari	0.0794
8.	Suspan	0.0377
9.	Kaitholia	0.0012
10.	Malihabad	0.0114
11.	Ater	0.0699
12.	Badkhorwa	0.0258
13.	Kamaaludi nnagar	0.0925
14.	Mundiyaana	0.0599
15.	Mahimakhera	0.0738

are applied, and look for alternatives where fertilizers might run off into surface waters. We have to preserve existing trees and plant new trees and shrubs to help prevent soil erosion and promote

infiltration of water into the soil. As we head into the 21st century, awareness and education will most assuredly continue to be the two most important ways to prevent water pollution. If these measures are not taken and water pollution continues, life on earth will suffer severely. But the developed world must work with the developing world to ensure that new industrialized economies do not add to the world's environmental problems. Conservation strategies need to be become more widely accepted and priority need to give to restore quality and quantity of aquifers before it is too.

ACKNOWLEDGEMENTS

The authors acknowledge the Head, Department of Environmental Sciences, Babasaheb Bhimrao Ambedkar (A Central) University Lucknow-226025 for providing facilities to this work. Support to Ms Anjali Verma in the form of Ph.D. fellowship is gratefully acknowledged.

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